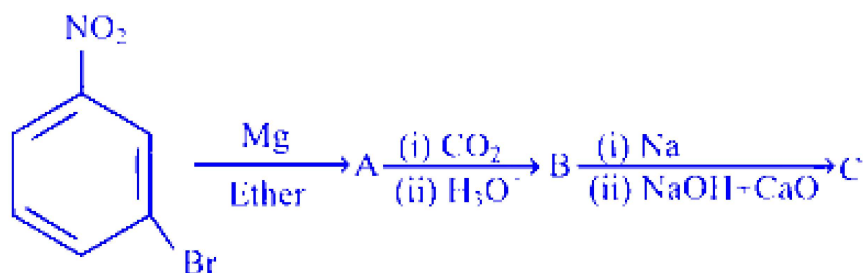


# Compounds Containing Nitrogen

## Question1

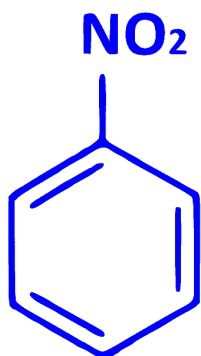
The product ' C ' in the given reaction sequence is



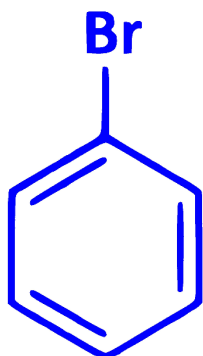
**TG EAPCET 2025 (Online) 2nd May Evening Shift**

Options:

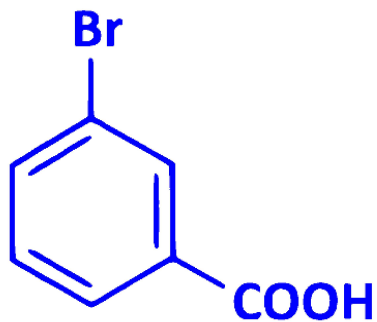
A.



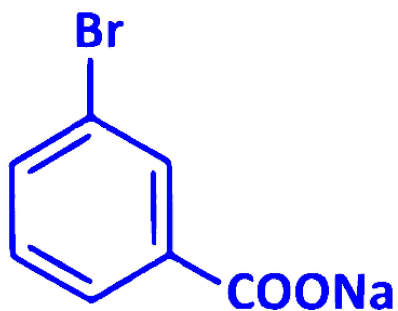
B.



C.



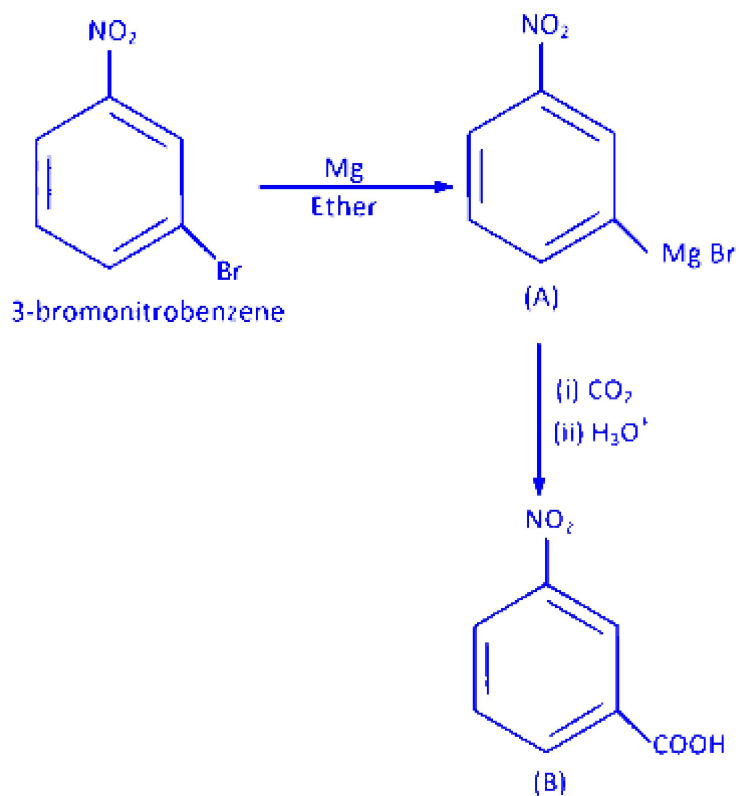
D.

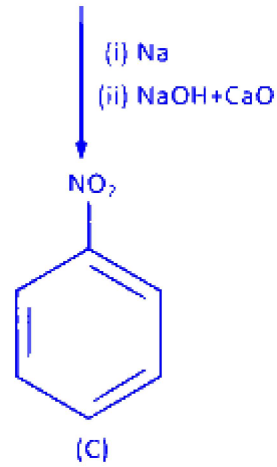


**Answer: A**

### Solution:

The complete reaction sequence involved is as follows





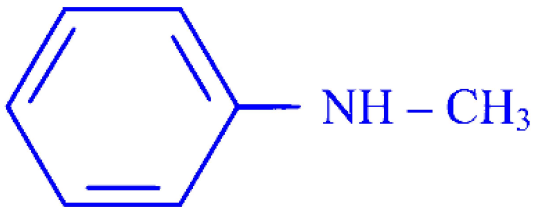
## Question2

The amine/salt of amine which gives positive test with a mixture of chloroform and alcoholic KOH solution is

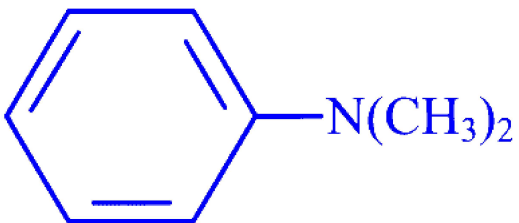
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Options:

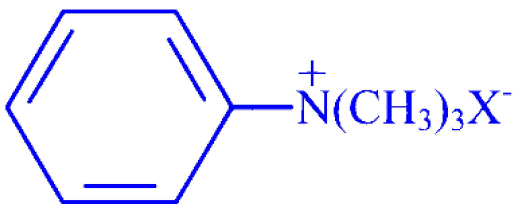
A.



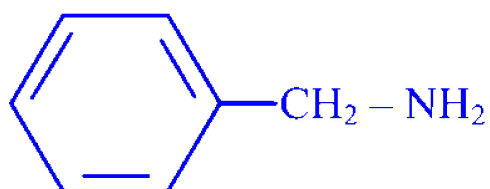
B.



C.



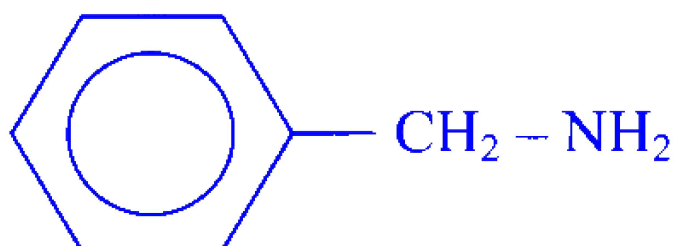
D.



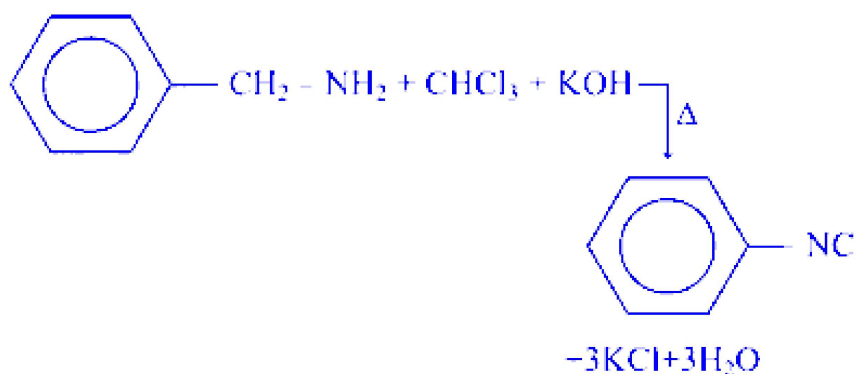
**Answer: D**

**Solution:**

Aliphatic and aromatic primary amines give positive carbylamine reaction. Thus, among the given options,



will give positive test with a mixture of chloroform and alc. KOH solution.



### Question3

The major products *P* and *Q* from the following reactions are

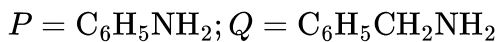


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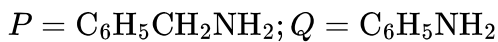


**Options:**

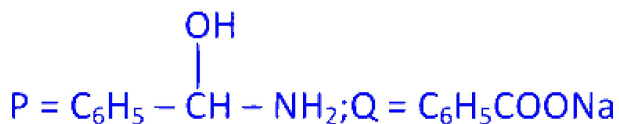
A.



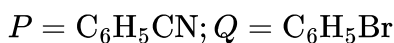
B.



C.



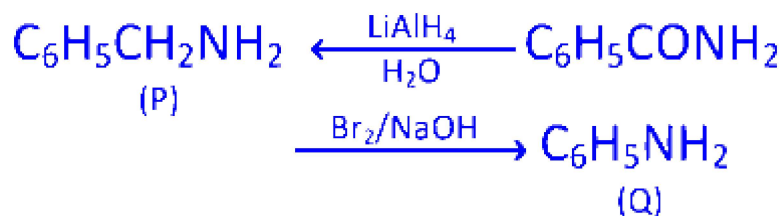
D.



**Answer: B**

**Solution:**

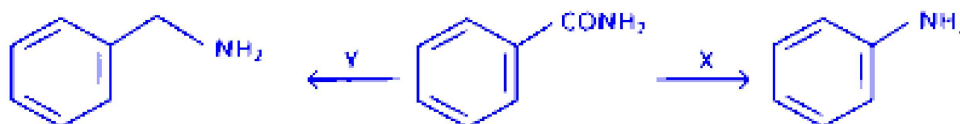
The complete reaction is as follows.



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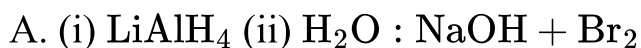
## Question4

What are X and Y respectively in the following reactions?



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**Options:**



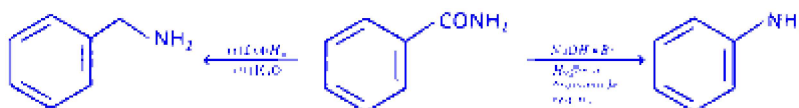
B.  $\text{NaOH} + \text{Br}_2$  : (i)  $\text{LiAlH}_4$  (ii)  $\text{H}_2\text{O}$

C.  $\text{NaOH} + \text{Br}_2$  : (i)  $\text{NaBH}_4$  (ii)  $\text{H}_2\text{O}$

D. (i)  $\text{NaBH}_4$  (ii)  $\text{H}_2\text{O}$  :  $\text{NaOH} + \text{Br}_2$

**Answer: B**

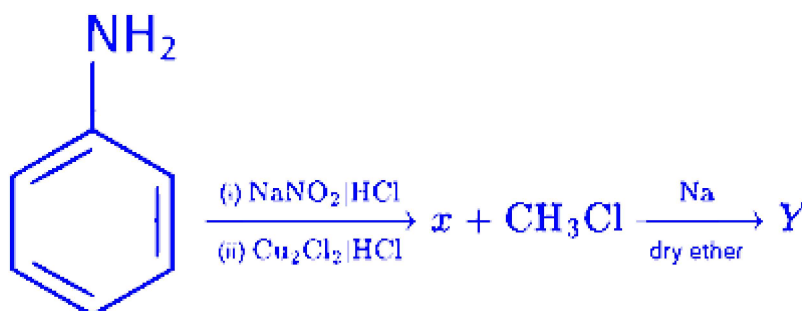
**Solution:**



Thus, X and Y are  $(\text{NaOH} + \text{Br}_2)$  and [(i)  $\text{LiAlH}_4$

(ii)  $\text{H}_2\text{O}$  ]

## Question 5



Conversion of X to Y in the above reaction is

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**Options:**

A. Wurtz reaction

B. Fittig reaction

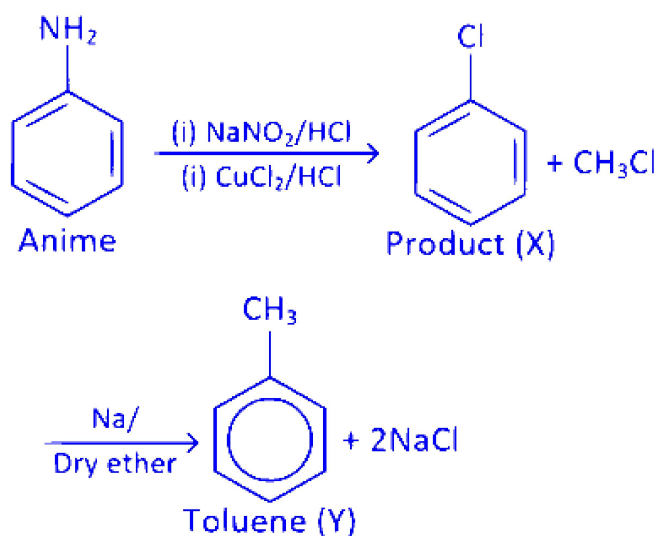
C. Wurtz-Fittig reaction

D. Sandmeyer reaction

**Answer: C**

## Solution:

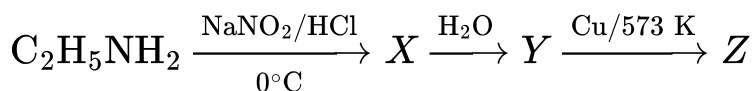
The Wurtz-Fittig reaction is a chemical process that produces substituted aromatic compound from aryl halides, alkyl halide in presence of dry ether and sodium metal. The complete reaction is as follows.



---

## Question 6

Identify the correct statements about Z.



- I. Z is an aldehyde.
- II. Z undergoes Cannizzaro reaction.
- III. Z gives iodoform test.
- IV. Z does not give test with Tollens' reagent.

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Options:

- A. I and III
- B. I and IV

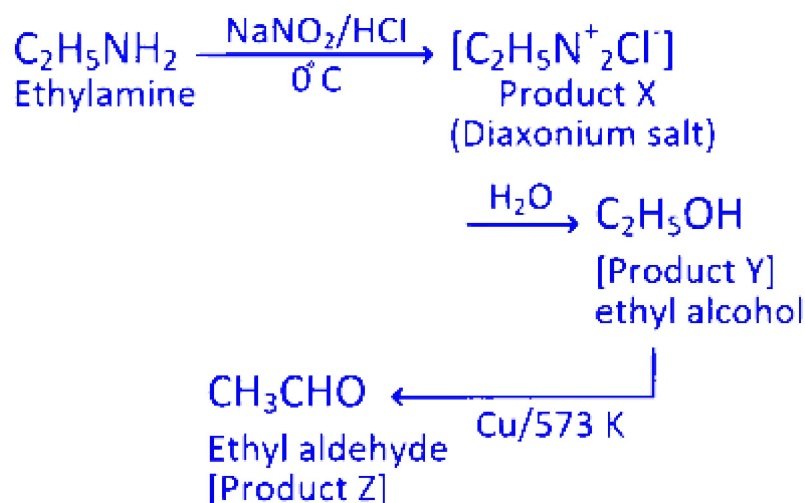
C. II and IV

D. II and III

**Answer: A**

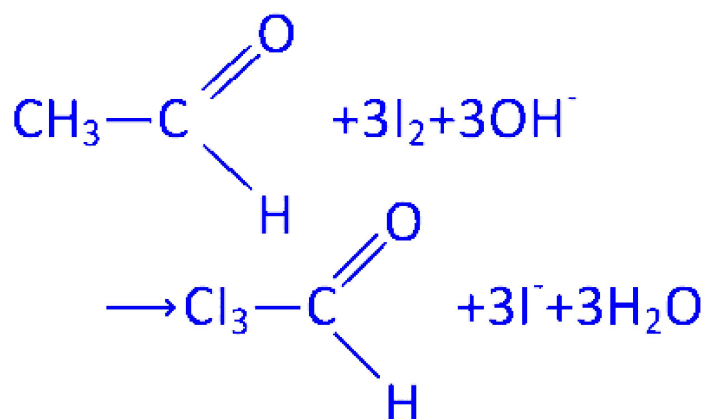
### Solution:

The complete reaction is as follows.



Product (Z) is an aldehyde.

Ethyl aldehyde is only aldehyde which gives iodoform test.



Hence, statements I and III are correct.

---

## Question7

### Match the following.

List -I (Amine)	List -II (pK <sub>b</sub> value)
A. N,N-dimethyl aniline	I. 9.30
B. Aniline	II. 8.92
C. N-ethylethanamine	III. 9.38
D. N-methylaniline	IV. 3.00

### The correct answer is

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#### Options:

- A. A-II, B - III, C - IV, D- I
- B. A-II, B - IV, C - III, D-I
- C. A-III, B -IV, C -I, D-II
- D. A-IV, B - III, C - II, D-I

**Answer: A**

#### Solution:

The lesser the pK<sub>b</sub> value, more stronger will be the base.

Hence, the correct match is A-II, B-III, C-IV, D-I.

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## Question8

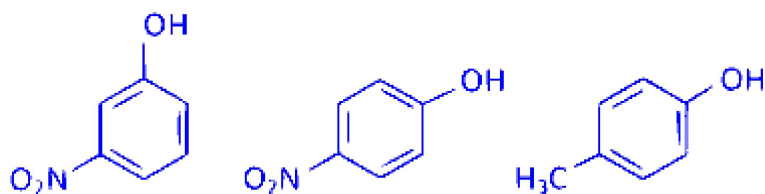
The pK<sub>a</sub> values of X, Y, Z respectively are 8.3, 7.1, 10.2. What are X, Y, Z ?

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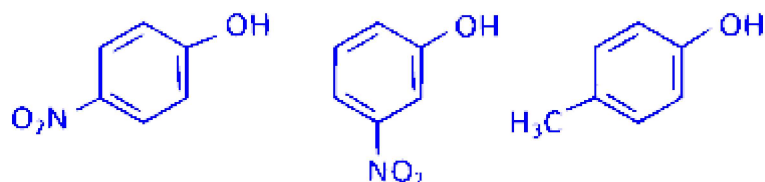


Options:

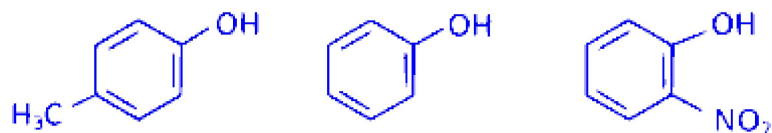
A.



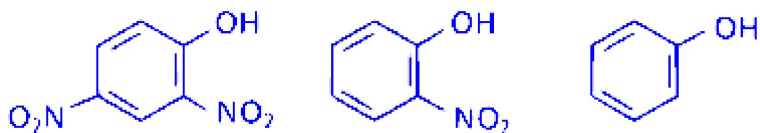
B.



C.



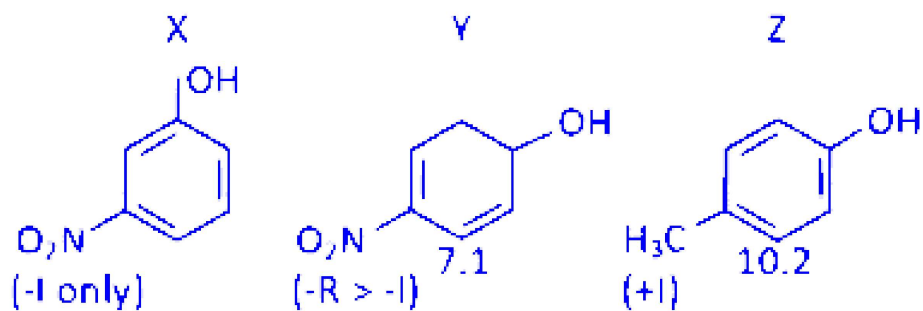
D.



**Answer: A**

**Solution:**

More is the  $pK_a$  value weaker is the acid. So, the compounds X, Y and Z are



## Question9

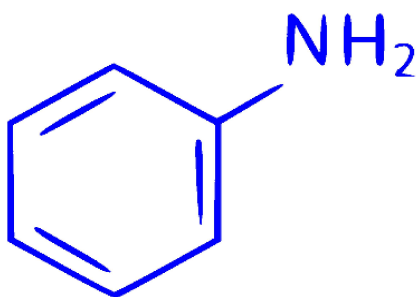
The reagents/ chemicals  $X$  and  $Y$  that convert cyanobenzene to Schiff's base are

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Options:

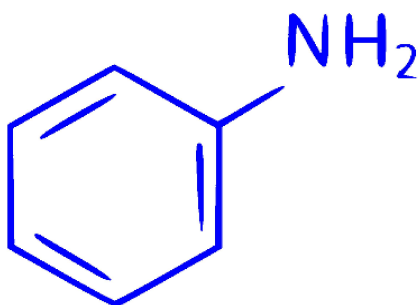
A. (i) DIBAL - H, (ii)  $\text{H}_2\text{O} \cdot \text{NH}_2\text{OH}$

B. DIBAL-  $\text{H}_1\text{H}_2\text{O}$ ,



C. LAH,  $\text{CH}_3\text{OH}$

D.  $\text{H}_3\text{O}^+$ ,

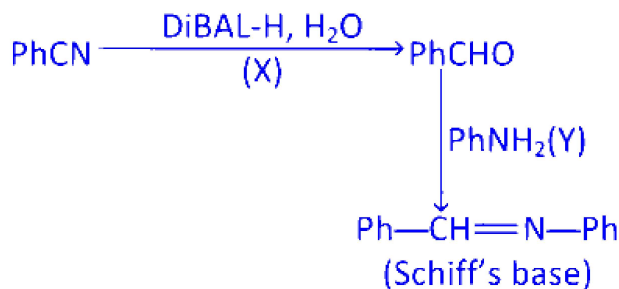


**Answer: B**

**Solution:**

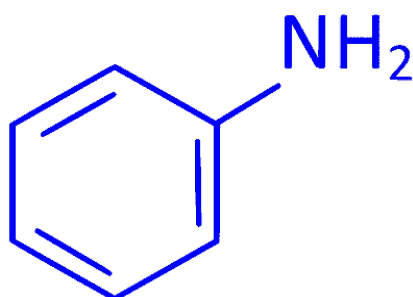
The reaction involved is as follows,





Thus, reagent/chemical *X* and *Y* are

DIBAL - H, H<sub>2</sub>O and



## Question10

The correct statement(s) of the following is/are (A) Aniline forms a stable benzene diazonium chloride at 285 K . (B) N - phenylethanamide is less reactive towards nitration than aniline. (C) *p* - CH<sub>3</sub>C<sub>6</sub>H<sub>4</sub>COCl is Hinsberg reagent.

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**Options:**

- A. A and B only
- B. A and C only
- C. B only
- D. C only

**Answer: C**



## Solution:

Statement given in B is correct which A and C are incorrect. The correct form of A and C are:

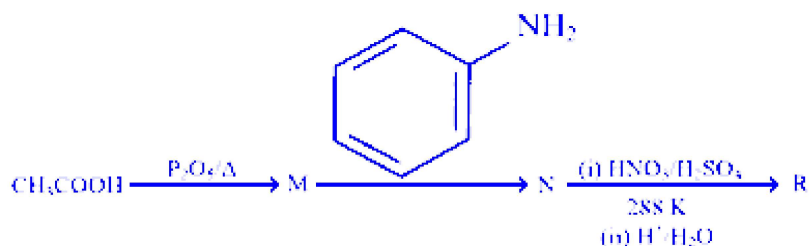
(A) Aniline can form stable diazonium salt at 273-283 K.

(C) Formula of Hinsberg reagent is  $C_6H_5SO_2Cl$ .

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## Question11

What is the major product 'R' in the following reaction sequence?



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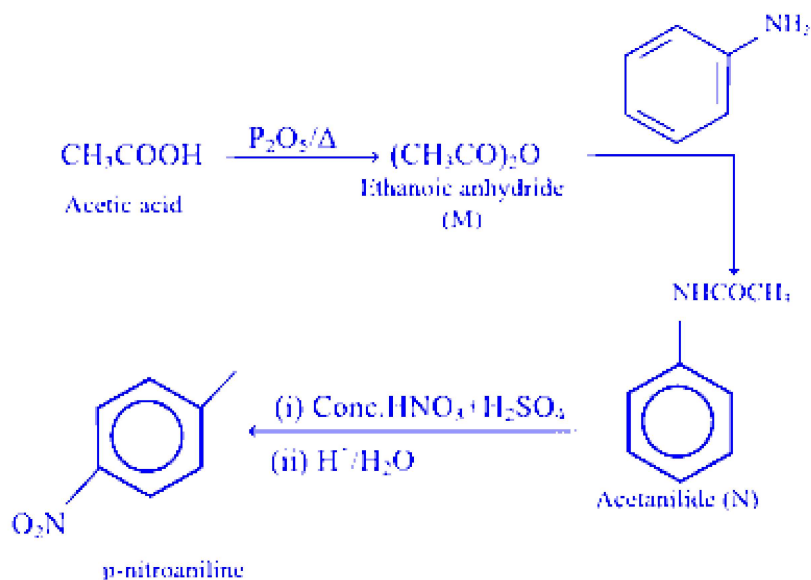
**Options:**

- A. o-nitro aniline
- B. m-nitro aniline
- C. p-nitroaniline
- D. p-aminobenzene sulphonic acid

**Answer: C**

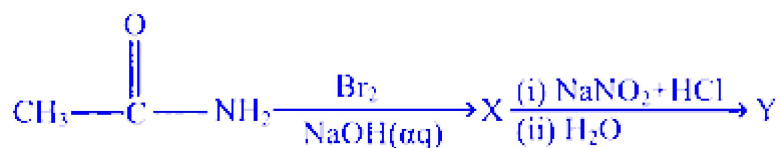
**Solution:**

Acetic acid is heated with  $P_2O_5$  to give ethanoic anhydride ( $M$ ).  $M$  reacts with aniline to give acetanilide  $N$ . At last, ( $N$ ) undergoes nitration to give  $p$ -nitroacetanilide which on acidic hydrolysis give  $p$ -nitroaniline ( $R$ ) as major product.



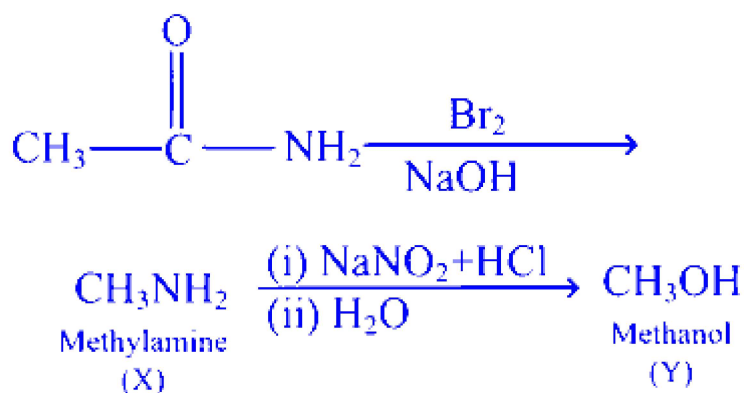
## Question 12

Identify what is  $Y$  in the following reaction sequence?



### Solution:

Acetamide reacts with  $\text{NaOH}$  and  $\text{Br}_2$  to give methyl amine ( $X$ ). In next step, methyl amine undergoes diazotisation reaction in presence of  $\text{NaNO}_2$  and  $\text{HCl}$ , which on further hydrolysis give methanol ( $Y$ ) as major product.



## Question13



Conversion of *A* to *B* is an example of the reaction

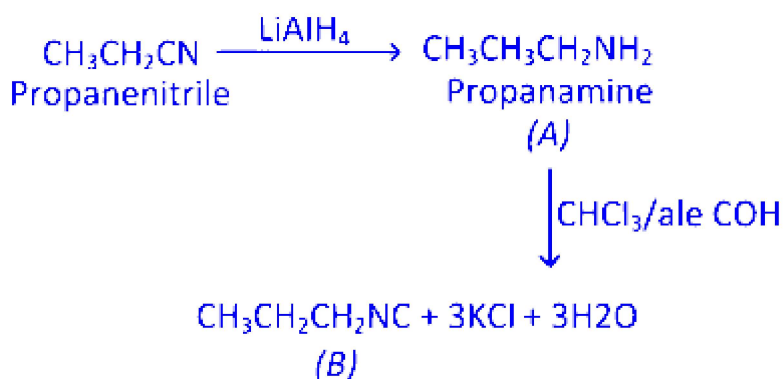
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Options:

- A. Reimer-Tiemann reaction
- B. Carbylamine reaction
- C. Stephen reaction
- D. Sandmeyer reaction

Answer: B

Solution:



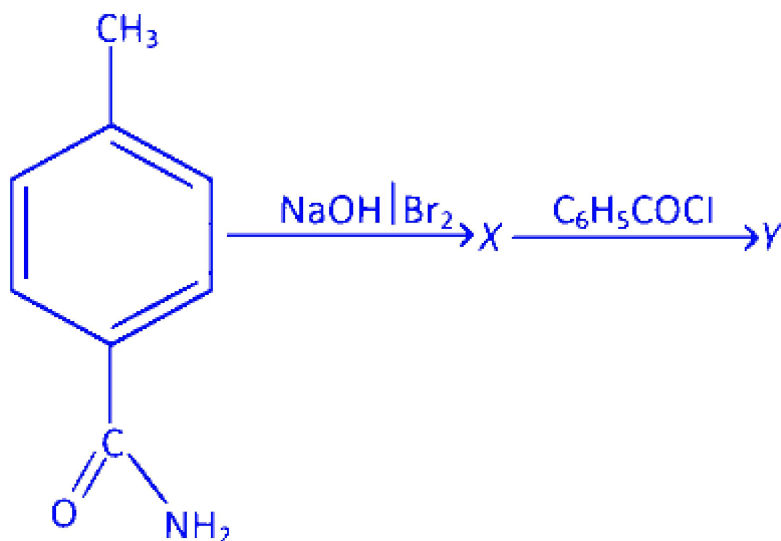
In the given reaction propanenitrile undergoes reduction on reaction with  $\text{LiAlH}_4$  to give primary amine *i.e.* propanamine (*A*) which reacts with  $\text{CHCl}_3$  in alc. KOH to give an isocyanide (*B*). This conversion of *A* to *B* is carbylamine reaction.

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## Question14

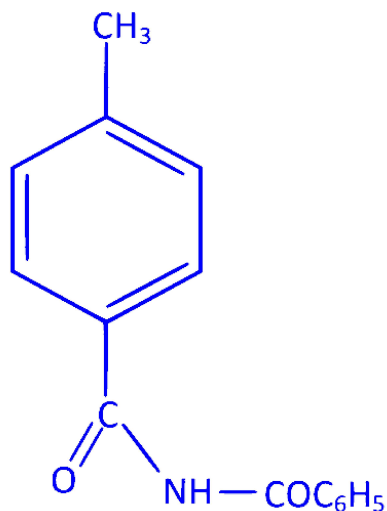
Identify the product ' Y ' in the following sequence of reactions



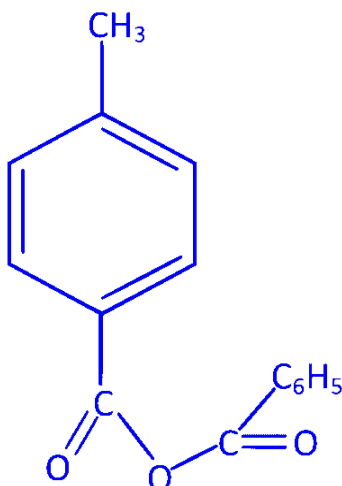
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Options:

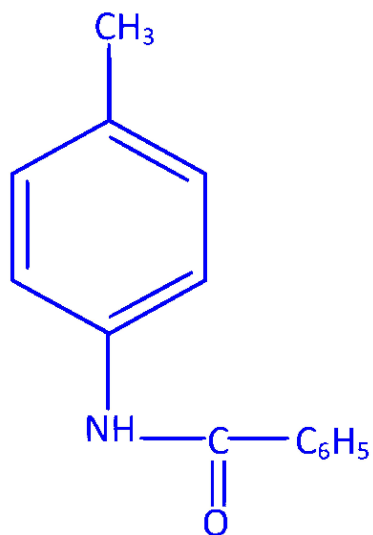
A.



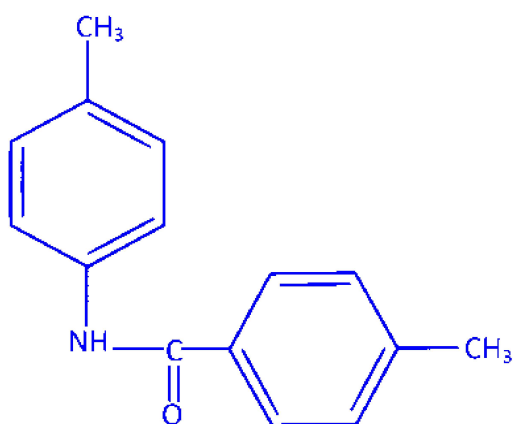
B.



C.



D.



**Answer: C**

**Solution:**

